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Chemical Kinetics - NCERT

www.ncerthelp.com (Visit for all ncert solutions in text and videos, CBSE syllabus, note and many more) Chemistry Notes for class 12 Chapter 4 Chemical Kinetics. The branch of chemistry, which deals with the rate of chemical reactions. the factors affecting the rate of reactions and the mechanism of the reaction. is called chemical kinetics. Chemical Reactions on the Basis of Rate of Reaction.

Chemistry Notes for class 12 Chapter 4 Chemical Kinetics

Chemical Kinetics is a branch of Chemistry which deals with chemical reaction, its factors and mechanism. It is closely related to the chemical reaction and physical process. Based on its varying rate, chemical kinetics Class 12 is divided into swift, prolonged and moderate reaction. Students learn various subtopics related to chemical kinetics like dependence on the rate of concentration, integrated equations, collision theory, catalyst, temperature dependence, and molecularity mechanism.

Chemical Kinetics NCERT Solutions - Class 12 Chemistry

Chemical Kinetics Class 12 Chemistry MCQs Pdf. 1. The half life period of first order reaction is 1386 seconds. The specific rate constant of the reaction is (a) 0.5 × 10⁻² s⁻¹ (b) 0.5 × 10⁻³ s⁻¹ (c) 5.0 × 10⁻² s⁻¹ (d) 5.0 × 10⁻³ s⁻¹. Answer/Explanation. Answer: b Explanation:

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Chemical Kinetics Class 12 Important Questions Short Answer Type - II [SA-II] Question 42. A first order reaction has a rate constant of 0.0051 min⁻¹. If we begin with 0.10 M concentration of the reactant, what concentration of reactant will remain in solution after 3 hours? (Delhi & All India 2009) Answer: Given : [R]₀ = 0.10 M, t = 3 hrs = 180 min

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Class 12 Chemistry Chapter 4 Chemical Kinetics This chapter deals with the kinetics, or the rate of a reaction. The topics covered include the factors affecting the rate of a reaction, the integrated rate equation, Pseudo First Order reactions, and the collision theory of chemical reactions.

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